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Reaction of Mes NPC1 with Triphenylcarbenium Tetrafluoroborate

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REACTION OF MES*NPCI WITH TRIPHENYLCARBENIUM TETRAFLUOROBORATE

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<u>Abstract</u> Reaction of Mes*NPCl with triphenylcarbenium salts (BF₄ or PF₆) produces a difluorophosphine and not the expected iminophosphenium cation. This compound then undergoes an Arbusov-type rearrangement to generate a difluoro-iminophosphorane.

Reaction of Mes*NPCl 1 with AlCl₃ or GaCl₃ gives the iminophosphenium cation, Mes*NP⁺. In contrast, reaction with NaBPh₄ gives the covalent iminophosphine Mes*NPPh¹ quantitatively. We now report the reaction with [Ph₃C][BF₄⁻] in 3:1 CH₂Cl₂/hexane solution (Scheme 1) which gives the (dialkyl)amino-difluorophosphine 2 quantitatively. This extremely sterically hindered phosphine then readily undergoes an Arbusov-type rearrangement to generate the difluoroiminophosphorane 3 which has been structurally characterized (Figure 1)². Compounds such as 3 typically dimerize, but this is prevented by steric hinderance³.

EXPERIMENTAL

Mes*NPCl (0.41 g, 1.26 mmol) in \approx 10 mL of 3:1 CH₂Cl₂/hexane was added over a period of 10 minutes to a solution of [Ph₃C][BF₄-] (0.42 g, 1.28 mmol) in \approx 10 mL of a similar solvent mixture. The solution turned bright yellow after 10 minutes and slow

removal of solvent yielded yellow crystals which were characterized as Mes*NPF₂CPh₃ (0.56 g, 0.98 mmol, 78%), mp 157.0-158.5 °C.

Elemental analysis: Calc. C, 77.73 H, 7.76 N, 2.45 Found C, 76.53 H, 8.34 N, 2.59

 31 P nmr studies of [Ph₃C][BF₄] or [Ph₃C][PF₆] indicate immediate quantitative formation of **2** (31 P: 146 ppm triplet; 1 J_{PF}: 1105 Hz) which then decays over three days to give **3** (31 P: -38 ppm triplet; 1 J _{PF}: 1221 Hz) in a solution yield of > 90%.

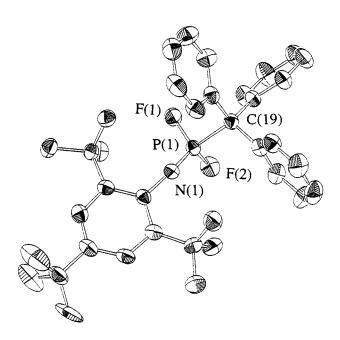


FIGURE 1 ORTEP view of <u>3</u>. Selected bond lengths (Å) and angles (°): P(1)-N(1) = 1.475(4), P(1)-C(19) = 1.847(4), P(1)-F(1) = 1.557(3), P(1)-F(2) = 1.562(3), $C_{Mes*}-N(1) = 1.563(3)$, $P(1)-N(1)-C_{Mes*} = 156.3(3)$.

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- 2. Crystal data $C_{37}H_{44}F_2NP$: M=571.73; space group P1 (#2), a = 13.728(3), b = 23.808(5), c = 10.125(2), $\alpha = 95.69(1)$, $\beta = 101.37(1)$, $\gamma = 91.62(2)$, V = 3224(1) Å³, Z = 2, $D_{calc.} = 1.178$ g cm⁻³, $\mu_{MoK\alpha} = 1.17$ cm⁻¹, Observations [I>3s(I)] 5430, 100R = 4.98, $100R_w = 4.98$, GoF = 1.532. Positional parameters have been deposited (CSD).
- 3. N. Burford, J. A. C. Clyburne, D. P. Gates, M. J. Schriver, and J. F. Richardson, J. Chem. Soc., Dalton Trans., 997 (1994), and references therein.